

Three-component Aza-Diels-Alder reactions using $\text{Yb}(\text{OTf})_3$ catalyst under conventional/ultrasonic techniques

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The $\text{Yb}(\text{OTf})_3$ catalyzed formal aza-Diels-Alder (or Povarov) reaction of cyclopentadiene and 1,3-cyclohexadiene with in situ-generated N-arylimines under conventional/ultrasonic techniques is herein described. This kind of three-component Povarov reaction results in quinoline and phenanthridine derivatives, which are important biological compounds.

Keywords: multi-component reaction, aza-Diels-Alder reaction, Povarov reaction, ultrasonic irradiation, lanthanide triflates, quinoline, phenanthridine

1. Introduction

Multi-component reactions allow the construction of several new bonds in a one-pot reaction. Therefore, multi-component reactions (MCRs) have increasing importance in synthetic organic chemistry and their methods provide significant benefits over conventional synthesis [1]. Among these benefits, a high degree of atomic economy, easier progress of reactions, decreased reaction times, low power consumption, and less handling of toxic substances are also the most important [2].

Quinoline and phenanthridine derivatives are one of the most important classes of natural products and exhibit a wide spectrum of biological activities such as psychotropic, cytotoxic, anti-allergic, anti-inflammatory, antipyretic, estrogenic, antimicrobial, antibacterial, antimalarial, antiplatelet, antitumoral and, anticancer, as insecticides, for the inhibition of HIV-1RT and as, dopamine D_1 receptor agonists [3]. The aza-Diels-Alder (or Povarov) reaction is a powerful and efficient means for the preparation of quinolines and phenanthridines [4]. The imines derived from aromatic amines and aldehydes act as heterodienes and react with various dienophiles in the presence of different acid catalysts. Lewis acids like AlCl_3 [5], ZnCl_2 [6], $\text{BF}_3 \cdot \text{OEt}_2$ [7], TiCl_4 [8], InCl_3 [9], ZrCl_4 [10], trifluoroacetic acid [11], p-TsOH [12], and HBF_4 [13] have been found to catalyze these reactions. However, Lewis acids often promote these reactions more than the stoichiometric amounts of the acids required because the acids can be trapped by the nitrogen of both the reactant and product. In recent years, lanthanide triflates [14], CAN [15], L-proline [16], protic acids [17], montmorillonite clay [18] and polymer-supported benzotriazole [19] have been reported to promote these reactions. Furthermore, lanthanide triflates are stable in water and only a

catalytic amount of the triflate is enough to complete these reactions. Moreover the aza-Diels-Alder reactions have been reported to proceed under photochemical conditions and microwave heating, as well as in aqueous media and ionic liquids [20].

Sonochemistry is the application of ultrasound to chemical reactions and processes. Luche and co-workers have written a number of studies that provided the fundamentals of sonochemistry [21]. In recent years, ultrasound irradiation has increasingly been used in organic synthesis [22]. This technique is able to activate many organic reactions due to cavitation collapse. Cavitation induces high temperatures and pressures inside the bubbles, leading to turbulent flow of the liquid and increased mass transfer [23]. Compared with traditional methods, this technique provides higher yields and selectivities, shorter reaction times and milder reaction conditions [24].

In this study, we have reported on the one-pot three-component, catalytic aza-Diels-Alder reaction using ytterbium (III) triflate (ytterbium (III) trifluoromethane sulfonate) under conventional/ultrasonic conditions.

2. Material and Methods

2.1 General Methods

NMR spectra and 2D experimental studies were determined on a Bruker Avance III-500 MHz NMR. Chemical shifts are given in ppm downfield from Me₄Si in CDCl₃ solution. Coupling constants are given in Hz. The FTIR spectra were recorded on a Perkin-Elmer FT-IR spectrometer (ATR) and absorption frequencies are reported in cm⁻¹. MS spectra were recorded on a Thermo Elemental X Series ICP-MS. Ultrasonication was performed in a Bandelin-Sonorex Ultrasonic Bath (Super RK) with a frequency of 35 kHz and a power of 230 W. The internal dimensions of the ultrasonic cleaner tank were 240x140x100 mm with liquid holding capacity of 3L. The reactor was a 100 mL pyrex round-bottom flask. The reaction flasks were suspended at the center of the bath, and the addition or removal of water controlled the temperature of the water bath. Melting points were measured on a Gallenkamp melting-point apparatus. Silica gel 60 (Merck) was used for column separation. TLC was conducted on standard conversion aluminum sheets pre-coated with a 0.2-mm layer of silica gel. All reagents were commercially available. Toluene was distilled and stored on sodium wire before use.

2.2 General Procedure for the Synthesis of Tetrahydroquinolines and Hexahydrophenanthridines under Conventional Conditions (**4a-p** and **6a-d**)

Aryl amine (1.00 mmol), aromatic aldehyde (1.00 mmol), freshly distilled cyclopentadiene or 1,3-cyclohexadiene (3.00 mmol) were added successively to a solution of Yb(OTf)₃ (62.2 mg, 0.1 mmol) in dry PhMe (5.00 mL) at room temperature. The reaction was refluxed at 110 °C for 6h under a nitrogen atmosphere. After completion of the reaction, as indicated by TLC monitoring, the reaction mixture was diluted with water (10 mL) and extracted with ethyl acetate (3x10 mL). The combined organic layers were dried over anhydrous Na₂SO₄, filtered and concentrated in a vacuum, and the resulting product was purified by column

chromatography over silica gel to afford the corresponding tetrahydroquinolines and hexahydrophenanthridines.

2.3 General Procedure for the Synthesis of Tetrahydroquinolines and Hexahydrophenanthridines under Ultrasonic Irradiation (**4a-p** and **6a-d**)

For the ultrasound-assisted method, a mixture of Yb(OTf)₃ (12.4 mg, 0.02 mmol), aryl amine (1.00 mmol), aromatic aldehyde (1.00 mmol) and freshly distilled cyclopentadiene or 1,3-cyclohexadiene (3.00 mmol) in dry PhMe (3.00 mL) was irradiated with ultrasound of low power (with a frequency of 35 kHz and a nominal power of 230W) at 50 °C under a nitrogen atmosphere for 40 minutes. The reaction flask was located at the maximum energy area in the cleaner and the surface of the reactants was placed slightly lower than the level of the water. The addition or removal of water controlled the temperature of the water bath. After completion of the reaction, as indicated by TLC monitoring, the reaction mixture was diluted with water (10 mL) and extracted with ethyl acetate (3x10 ml). The combined organic layers were dried over anhydrous Na₂SO₄, filtered and concentrated in a vacuum. The resulting product was purified by column chromatography over silica gel to afford the corresponding tetrahydroquinolines and hexahydrophenanthridines.

2.3.1 *cis*-6-phenyl-6,6a,7,9a-tetrahydro-5H-benzo[*h*]cyclopenta[*c*]quinoline (**4a**)

Pale white crystals, m.p. 105-107 °C. FTIR (ATR, cm⁻¹): 3377, 3053, 3024, 2914, 2848, 1602, 1573, 1516, 1494, 1460, 1401, 1375, 1355, 1291, 1273, 1240, 1141, 1095, 1088, 1029, 813, 788, 751, 729. ¹H NMR (CDCl₃, 500 MHz): 1.86-1.91 (dd, *J*₁ = 8.50 Hz, *J*₂ = 7.50 Hz, 1H), 2.72-2.77 (dt, *J*₁ = 8.50 Hz, *J*₂ = 2.00 Hz, 1H), 3.10-3.15 (ddd, *J*₁ = 8.50 Hz, *J*₂ = 6.00 Hz, *J*₃ = 2.00 Hz, 1H), 4.28-4.29 (d, *J* = 8.50 Hz, 1H), 4.50 (br s, 1H), 4.75-4.76 (d, *J* = 2.00 Hz, 1H), 5.66 (br s, 1H), 5.94 (br s, 1H), 7.21-7.23 (d, *J* = 8.00 Hz, 1H), 7.28-7.30 (d, *J* = 8.50 Hz, 1H), 7.33-7.47 (m, 5H), 7.56-7.58 (d, *J* = 8.00 Hz, 2H), 7.75-7.77 (d, *J* = 8.00 Hz, 2H). ¹³C NMR (CDCl₃, 125 MHz): 31.35, 46.17, 47.05, 58.25, 118.77, 119.52, 120.21, 123.66, 125.08, 125.11, 126.71, 127.35, 127.38, 128.63, 130.67, 132.59, 134.02, 139.83, 143.07. MS *m/z* (ESI): 298 (M⁺), 245, 232, 211, 195, 172.

2.3.2 *cis*-6-(4-bromophenyl)-6,6a,7,9a-tetrahydro-5H-benzo[*h*]cyclopenta[*c*]quinoline (**4b**)

Brown crystals, m.p. 170-172 °C. FTIR (ATR, cm⁻¹): 3372, 3050, 2912, 2847, 1580, 1575, 1516, 1486, 1461, 1394, 1355, 1292, 1240, 1195, 1141, 1070, 1008, 812, 789, 752, 728. ¹H NMR (CDCl₃, 500 MHz): 1.84-1.88 (dd, *J*₁ = 8.50 Hz, *J*₂ = 7.50 Hz, 1H), 2.65-2.70 (dt, *J*₁ = 8.50 Hz, *J*₂ = 2.50 Hz, 1H), 3.04-3.10 (ddd, *J*₁ = 8.50 Hz, *J*₂ = 6.00 Hz, *J*₃ = 2.50 Hz, 1H), 4.25-4.27 (d, *J* = 8.50 Hz, 1H), 4.41 (br s, 1H), 4.70-4.71 (d, *J* = 2.50 Hz, 1H), 5.65 (br s, 1H), 5.94 (br s, 1H), 7.20-7.22 (d, *J* = 8.50 Hz, 1H), 7.29-7.31 (d, *J* = 8.00 Hz, 1H), 7.39-7.45 (m, 4H), 7.54-7.57 (d, *J* = 8.00 Hz, 2H), 7.74-7.77 (t, *J* = 8.50 Hz, 2H). ¹³C NMR (CDCl₃, 125 MHz): 31.23, 45.96, 46.89, 57.72, 119.06, 119.43, 120.17, 121.10, 123.65, 125.18, 127.26, 128.40, 128.66, 130.54, 131.71, 132.56, 133.98, 139.42, 142.08. MS *m/z* (ESI): 376 (M⁺), 311, 221, 205, 168.

2.3.3 *cis*-6-(4-chlorophenyl)-6,6a,7,9a-tetrahydro-5H-benzo[*h*]cyclopenta[*c*]quinoline (**4c**)

Pale white crystals, m.p. 166-168 °C. FTIR (ATR, cm^{-1}): 3395, 3062, 3024, 2926, 2908, 2845, 1570, 1517, 1487, 1458, 1395, 1372, 1353, 1292, 1270, 1238, 1136, 1085, 1009, 815, 789, 772, 753, 737. ^1H NMR (CDCl_3 , 500 MHz): 1.84-1.89 (dd, $J_1 = 8.50$ Hz, $J_2 = 7.50$ Hz, 1H), 2.66-2.71 (dt, $J_1 = 8.50$ Hz, $J_2 = 3.00$ Hz, 1H), 3.04-3.10 (ddd, $J_1 = 8.50$ Hz, $J_2 = 6.00$ Hz, $J_3 = 3.00$ Hz, 1H), 4.26-4.28 (d, $J = 8.50$ Hz, 1H), 4.42 (br s, 1H), 4.73 (d, $J = 3.00$ Hz, 1H), 5.65 (br s, 1H), 5.95 (br s, 1H), 7.19-7.23 (d, $J = 8.00$ Hz, 1H), 7.28-7.33 (d, $J = 8.00$ Hz, 1H), 7.39-7.45 (m, 4H), 7.50-7.52 (d, $J = 8.00$ Hz, 2H), 7.74-7.77 (t, $J = 7.00$ Hz, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.26, 46.05, 46.93, 57.70, 119.06, 119.46, 120.20, 123.69, 125.20, 127.29, 128.06, 128.68, 128.79, 130.57, 132.60, 133.06, 134.02, 139.48, 141.58. MS m/z (ESI): 332 (M⁺), 293, 266, 245, 216, 194, 172.

2.3.4 *trans*-6-(4-chlorophenyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (**4c'**)

pale yellow oil, ^1H NMR (CDCl_3 , 500 MHz): 2.09-2.13 (d, $J = 16.50$ Hz, 1H), 2.50-2.54 (m, 1H), 2.73-2.78 (dd, $J_1 = 10.50$ Hz, $J_2 = 7.00$ Hz, 1H), 3.80-3.82 (d, $J = 10.50$ Hz, 1H), 4.11 (br s, 1H), 4.55 (br s, 1H), 5.72 (br s, 1H), 6.05 (br s, 1H), 7.31-7.34 (m, 2H), 7.37-7.50 (m, 6H), 7.62-7.64 (d, $J = 8.00$ Hz, 1H), 7.77-7.83 (m, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 35.78, 42.81, 47.57, 57.80, 118.20, 119.43, 119.81, 122.96, 125.24, 127.75, 128.25, 128.63, 128.80, 130.01, 132.71, 133.72, 136.21, 139.83, 141.45

2.3.5 *cis*-6-(*p*-tolyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (**4d**)

Pale white crystals, m.p. 122-124 °C. FTIR (ATR, cm^{-1}): 3376, 3051, 3020, 2918, 2870, 2849, 1611, 1600, 1573, 1514, 1495, 1459, 1398, 1375, 1356, 1291, 1240, 1195, 1141, 1107, 1032, 1020, 999, 811, 787, 753, 728. ^1H NMR (CDCl_3 , 500 MHz): 1.80-1.85 (dd, $J_1 = 8.50$ Hz, $J_2 = 7.50$ Hz, 1H), 2.32 (s, 3H), 2.63-2.69 (dt, $J_1 = 8.50$ Hz, $J_2 = 3.00$ Hz, 1H), 2.99-3.05 (ddd, $J_1 = 8.50$ Hz, $J_2 = 6.00$ Hz, $J_3 = 3.00$ Hz, 1H), 4.18-4.20 (d, $J = 8.50$ Hz, 1H), 4.39 (br s, 1H), 4.63-4.64 (d, $J = 3.00$ Hz, 1H), 5.58 (br s, 1H), 5.86 (br s, 1H), 7.12-7.14 (d, $J = 8.50$ Hz, 1H), 7.15-7.17 (m, 2H), 7.19-7.21 (d, $J = 8.50$ Hz, 1H), 7.31-7.35 (m, 2H), 7.36-7.38 (d, $J = 8.50$ Hz, 2H), 7.66-7.68 (t, $J = 7.00$ Hz, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 20.10, 30.34, 45.16, 46.04, 57.00, 117.63, 118.49, 119.18, 122.60, 124.00, 125.58, 126.33, 127.58, 128.24, 129.66, 131.54, 133.01, 135.97, 138.92, 139.01. MS m/z (ESI): 311 (M⁺), 246, 220, 206, 169.

2.3.6 *cis*-6-(4-methoxyphenyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (**4e**)

White crystals, m.p. 129-131 °C. FTIR (ATR, cm^{-1}): 3377, 3051, 2997, 2948, 2928, 2834, 1610, 1573, 1510, 1459, 1399, 1375, 1355, 1293, 1241, 1172, 1141, 1107, 1032, 839, 813, 788, 753, 732. ^1H NMR (CDCl_3 , 500 MHz): 1.88-1.93 (dd, $J_1 = 8.50$ Hz, $J_2 = 7.50$ Hz, 1H), 2.70-2.76 (dt, $J_1 = 8.50$ Hz, $J_2 = 3.00$ Hz, 1H), 3.04-3.10 (ddd, $J_1 = 8.50$ Hz, $J_2 = 6.50$ Hz, $J_3 = 3.00$ Hz, 1H), 3.85 (s, 3H), 4.24-4.26 (d, $J = 8.50$ Hz, 1H), 4.43 (s, 1H), 4.69-4.70 (d, $J = 3.00$ Hz, 1H), 5.66 (br s, 1H), 5.94 (br s, 1H), 6.95-6.97 (d, $J = 8.50$ Hz, 2H), 7.20-7.22 (d, $J = 8.00$ Hz, 1H), 7.27-7.29 (d, $J = 8.50$ Hz, 1H), 7.37-7.42 (m, 2H), 7.47-7.49 (d, $J = 8.50$ Hz, 2H), 7.73-7.76 (m, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.40, 46.32, 47.07, 55.38, 57.73, 113.99, 118.71, 119.57, 120.22, 123.66, 125.06, 125.11, 127.40, 127.80, 128.64, 130.72, 132.61, 134.10, 135.21, 139.99, 158.94. MS m/z (ESI): 327 (M⁺), 263, 221, 205, 169.

2.3.7 *cis-4-(6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinolin-6-yl)phenol (4f)*

Pale yellow crystals, m.p. 80-82 °C. FTIR (ATR, cm^{-1}): 3367, 3048, 3011, 2926, 2868, 1600, 1580, 1569, 1513, 1444, 1403, 1375, 1355, 1285, 1235, 1216, 1160, 1100, 1082, 1040, 1014, 833, 789, 769. ^1H NMR (CDCl_3 , 500 MHz): 1.88-1.93 (dd, $J_1=8.50$ Hz, $J_2=7.50$ Hz, 1H), 2.70-2.75 (dt, $J_1=8.50$ Hz, $J_2=3.00$ Hz, 1H), 3.04-3.10 (ddd, $J_1=8.50$ Hz, $J_2=6.00$ Hz, $J_3=3.00$ Hz, 1H), 4.25-4.27 (d, $J=8.50$ Hz, 1H), 4.43 (br s, 1H), 4.69-4.70 (d, $J=3.00$ Hz, 1H), 4.91 (br s, 1H), 5.67 (br s, 1H), 5.94 (br s, 1H), 6.88-6.90 (d, $J=8.50$ Hz, 2H), 7.20-7.22 (d, $J=8.50$ Hz, 1H), 7.27-7.29 (d, $J=8.50$ Hz, 1H), 7.39-7.44 (m, 4H), 7.74-7.76 (m, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.36, 46.30, 47.04, 57.72, 115.41, 118.73, 119.55, 120.20, 123.65, 125.05, 125.10, 127.37, 128.00, 128.63, 130.69, 132.60, 134.08, 135.38, 139.93, 154.85. MS m/z (ESI): 313 (M^+), 248, 220, 205, 169.

2.3.8 *cis-4-(6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinolin-6-yl)benzotrile (4g)*

Pale white crystals, m.p. 208-210 °C. FTIR (ATR, cm^{-1}): 3361, 3044, 2961, 2931, 2905, 2844, 2226, 1606, 1573, 1521, 1477, 1439, 1400, 1373, 1357, 1338, 1292, 1242, 1194, 1152, 1094, 1031, 813, 790, 752, 735. ^1H NMR (CDCl_3 , 500 MHz): 1.71-1.76 (dd, $J_1=8.50$ Hz, $J_2=7.00$ Hz, 1H), 2.55-2.60 (dt, $J_1=8.50$ Hz, $J_2=3.00$ Hz, 1H), 2.99-3.05 (ddd, $J_1=8.50$ Hz, $J_2=6.00$ Hz, $J_3=3.00$ Hz, 1H), 4.20-4.22 (d, $J=8.50$ Hz, 1H), 4.35 (br s, 1H), 4.72 (br s, 1H), 5.57 (br s, 1H), 5.88 (br s, 1H), 7.13-7.15 (d, $J=8.50$ Hz, 1H), 7.24-7.26 (d, $J=8.50$ Hz, 1H), 7.33-7.39 (m, 2H), 7.60-7.62 (d, $J=8.00$ Hz, 2H), 7.64-7.65 (d, $J=8.00$ Hz, 2H), 7.67-7.71 (t, $J=8.00$ Hz, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.16, 45.77, 46.82, 58.03, 111.29, 118.82, 119.35, 119.46, 120.15, 123.71, 125.32, 127.17, 127.46, 128.73, 130.37, 132.50, 132.58, 133.96, 138.90, 148.55. MS m/z (ESI): 323 (M^+), 258, 221, 206, 170.

2.3.9 *cis-6-(4-nitrophenyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (4h)*

Pale red crystals, m.p. 212-214 °C. FTIR: 3376, 3100, 3077, 3055, 2929, 2907, 2873, 2846, 1596, 1573, 1509, 1476, 1402, 1342, 1293, 1241, 1150, 1108, 1034, 1013, 860, 817, 791, 771, 751, 723. ^1H NMR (CDCl_3 , 500 MHz): 1.71-1.76 (dd, $J_1=8.50$ Hz, $J_2=6.50$ Hz, 1H), 2.57-2.62 (dt, $J_1=8.50$ Hz, $J_2=2.50$ Hz, 1H), 3.01-3.07 (ddd, $J_1=8.50$ Hz, $J_2=6.00$ Hz, $J_3=2.50$ Hz, 1H), 4.22-4.23 (d, $J=8.50$ Hz, 1H), 4.38 (br s, 1H), 4.77-4.78 (d, $J=2.50$ Hz, 1H), 5.58 (br s, 1H), 5.89 (br s, 1H), 7.14-7.16 (d, $J=8.50$ Hz, 1H), 7.18 (br s, 1H), 7.25-7.27 (d, $J=8.00$ Hz, 1H), 7.34-7.40 (m, 2H), 7.66-7.68 (d, $J=8.50$ Hz, 2H), 7.69-7.71 (d, $J=8.00$ Hz, 2H), 8.20-8.22 (d, $J=8.50$ Hz, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.19, 45.77, 46.84, 57.90, 119.38, 119.60, 120.18, 123.74, 123.95, 125.36, 125.41, 127.18, 127.51, 128.77, 130.37, 132.61, 133.99, 138.84, 147.33, 150.60. MS m/z (ESI): 343 (M^+), 297, 277, 231, 173.

2.3.10 *cis-6-(2-nitrophenyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (4i)*

Pale yellow crystals, m.p. 106-108 °C. FTIR: 3372, 3060, 3051, 2921, 2850, 1620, 1608, 1571, 1515, 1470, 1440, 1392, 1337, 1310, 1285, 1262, 1148, 1082, 1039, 1015, 802, 773, 753, 724. ^1H NMR (CDCl_3 , 500 MHz): 1.83-1.87 (dd, $J_1=8.50$ Hz, $J_2=6.00$ Hz, 1H), 2.73-2.78 (dt, $J_1=8.50$ Hz, $J_2=2.50$ Hz, 1H), 3.35-3.40 (ddd, $J_1=8.50$ Hz, $J_2=6.00$ Hz, $J_3=2.50$ Hz, 1H), 4.27 (br s, 1H), 4.29-4.31 (d, $J=8.50$ Hz, 1H), 5.24-5.25 (d, $J=2.50$ Hz, 1H), 5.66 (br

s, 1H), 5.98 (br s, 1H), 7.15-7.16 (d, $J = 7.50$ Hz, 1H), 7.47-7.50 (t, $J = 7.50$ Hz, 1H), 7.52-7.54 (m, 2H), 7.61-7.64 (t, $J = 7.00$ Hz, 1H), 7.75-7.78 (m, 2H), 7.85-7.87 (m, 1H), 8.06-8.08 (d, $J = 8.00$ Hz, 1H), 8.34-8.36 (m, 1H), 8.46-8.47 (d, $J = 8.00$ Hz, 1H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.14, 43.79, 46.78, 53.18, 113.25, 123.67, 124.55, 126.01, 126.04, 126.53, 126.85, 127.75, 128.77, 130.01, 131.12, 131.26, 133.54, 133.90, 148.20, 155.80. MS m/z (ESI): 343 (M^+), 297, 277, 231, 173.

2.3.11 *cis*-6-(4-pyridin-2-yl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (**4j**)

Pale orange oil. FTIR: 3377, 3050, 3006, 2919, 2847, 1619, 1591, 1574, 1514, 1453, 1434, 1401, 1374, 1354, 1287, 1240, 1196, 1144, 1098, 1047, 1031, 813, 788, 770, 752, 733. ^1H NMR (CDCl_3 , 500 MHz): 1.90-1.94 (dd, $J_1 = 8.50$ Hz, $J_2 = 7.00$ Hz, 1H), 2.46-2.50 (dt, $J_1 = 8.50$ Hz, $J_2 = 2.50$ Hz, 1H), 3.41-3.46 (ddd, $J_1 = 8.50$ Hz, $J_2 = 6.00$ Hz, $J_3 = 2.50$ Hz, 1H), 4.30-4.32 (d, $J = 8.50$ Hz, 1H), 4.81-4.82 (d, $J = 2.50$ Hz, 1H), 5.47 (br s, 1H), 5.61 (br s, 1H), 5.91 (br s, 1H), 7.18-7.20 (d, $J = 7.50$ Hz, 1H), 7.26-7.30 (m, 2H), 7.37-7.44 (m, 3H), 7.48-7.50 (d, $J = 7.50$ Hz, 1H), 7.68-7.78 (m, 3H), 7.90-7.92 (d, $J = 8.50$ Hz, 1H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.56, 44.12, 47.10, 57.99, 109.64, 118.53, 118.91, 120.09, 120.46, 122.11, 124.92, 125.10, 127.22, 128.45, 130.16, 132.64, 134.25, 136.59, 139.25, 148.93, 160.59. MS m/z (ESI): 299 (M^+), 233, 220, 206, 156.

2.3.12 *cis*-6-(thiophen-2-yl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (**4k**)

Pale orange oil. FTIR (ATR, cm^{-1}): 3287, 3080, 3010, 2970, 2882, 1610, 1589, 1574, 1518, 1458, 1424, 1401, 1378, 1348, 1290, 1240, 1191, 1134, 1092, 1041, 1019, 823, 768, 751. ^1H NMR (CDCl_3 , 500 MHz): 2.11-2.16 (dd, $J_1 = 8.00$ Hz, $J_2 = 7.50$ Hz, 1H), 2.78-2.84 (dt, $J_1 = 8.00$ Hz, $J_2 = 3.00$ Hz, 1H), 3.10-3.16 (ddd, $J_1 = 8.00$ Hz, $J_2 = 6.00$ Hz, $J_3 = 3.00$ Hz, 1H), 4.22-4.24 (d, $J = 8.00$ Hz, 1H), 4.53 (br s, 1H), 4.99-5.00 (d, $J = 3.00$ Hz, 1H), 5.67 (br s, 1H), 5.92 (br s, 1H), 7.03-7.05 (t, $J = 3.50$ Hz, 1H), 7.17-7.19 (d, $J = 9.00$ Hz, 2H), 7.26-7.29 (m, 2H), 7.36-7.41 (m, 2H), 7.70-7.74 (t, $J = 9.00$ Hz, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 32.04, 46.20, 46.86, 54.67, 119.18, 119.58, 120.47, 123.62, 123.83, 124.13, 125.14, 126.67, 127.21, 128.20, 128.53, 129.01, 130.63, 132.48, 133.92, 139.20, 146.34. MS m/z (ESI): 304 (M^+), 239, 221, 206, 170.

2.3.13 *cis*-6-(furan-3-yl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (**4l**)

Pale yellow oil. FTIR (ATR, cm^{-1}): 3385, 3090, 3054, 2948, 2926, 2869, 2855, 1574, 1517, 1463, 1403, 1387, 1356, 1293, 1270, 1240, 1161, 1140, 1096, 1067, 1026, 873, 788, 752, 730. ^1H NMR (CDCl_3 , 500 MHz): 2.12-2.17 (dd, $J_1 = 8.50$ Hz, $J_2 = 6.00$ Hz, 1H), 2.71-2.76 (dt, $J_1 = 8.50$ Hz, $J_2 = 2.50$ Hz, 1H), 3.06-3.12 (ddd, $J_1 = 8.50$ Hz, $J_2 = 6.00$ Hz, $J_3 = 2.50$ Hz, 1H), 4.21-4.23 (d, $J = 8.50$ Hz, 1H), 4.37 (br s, 1H), 4.65-4.66 (d, $J = 2.50$ Hz, 1H), 5.68 (br s, 1H), 5.91 (br s, 1H), 6.53 (br s, 1H), 7.18-7.20 (d, $J = 8.50$ Hz, 1H), 7.27-7.29 (d, $J = 8.50$ Hz, 1H), 7.37-7.43 (m, 3H), 7.46-7.47 (d, $J = 3$ Hz, 1H), 7.56-7.57 (d, $J = 3$ Hz, 1H), 7.70-7.75 (m, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 32.11, 44.85, 46.81, 50.95, 109.53, 114.97, 118.96, 119.53, 120.55, 123.64, 125.12, 127.37, 128.62, 130.63, 132.55, 134.19, 139.19, 143.25. MS m/z (ESI): 288 (M^+), 223, 192, 156.

4.3.14 *cis*-6-(5-methylfuran-2-yl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (4m)

Pale yellow oil. FTIR (ATR, cm^{-1}): 3350, 3060, 3012, 2967, 2864, 1621, 1600, 1584, 1519, 1468, 1444, 1403, 1362, 1343, 1270, 1238, 1181, 1142, 1083, 1042, 1012, 943, 834, 768, 736. ^1H NMR (CDCl_3 , 500 MHz): 2.14-2.18 (dd, $J_1=8.50$ Hz, $J_2=6.50$ Hz, 1H), 2.22 (s, 3H), 2.64-2.67 (dt, $J_1=8.50$ Hz, $J_2=2.50$ Hz, 1H), 3.12-3.17 (ddd, $J_1=8.50$ Hz, $J_2=6.00$ Hz, $J_3=2.50$ Hz, 1H), 4.09-4.11 (d, $J=8.50$ Hz, 1H), 4.45 (br s, 1H), 4.56 (br s), 5.86 (br s, 1H), 6.11 (br s), 7.06-7.08 (d, $J=8.50$ Hz, 1H), 7.15-7.17 (d, $J=8.50$ Hz, 1H), 7.26-7.30 (t, $J=6.50$ Hz, 4H), 7.62-7.64 (t, $J=6.50$ Hz, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 13.74, 32.53, 42.77, 47.00, 52.86, 106.24, 106.46, 119.08, 119.77, 120.73, 123.78, 125.16, 125.23, 127.43, 128.67, 130.77, 132.63, 134.13, 139.49, 151.38, 154.05. MS m/z (ESI): 302 (M^+), 237, 221, 206, 170.

2.3.15 *trans*-6-(5-methylfuran-2-yl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (4m')

Pale yellow oil. ^1H NMR (CDCl_3 , 500 MHz): 2.25-2.28 (m, 1H), 2.45 (s, 3H), 2.71-2.80 (m, 1H), 3.20-3.26 (m, 1H), 3.46 (br s), 4.25-4.27 (d, $J=16.00$ Hz, 1H), 4.65 (br s, 1H), 5.96 (br s, 1H), 6.20 (br s, 1H), 7.05 (br s, 1H), 7.11-7.12 (d, $J=3.00$ Hz, 1H), 7.37-7.42 (m, 2H), 7.67-7.72 (m, 2H), 7.82-7.85 (t, $J=7.50$ Hz, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 14.14, 32.41, 42.64, 46.85, 50.92, 106.36, 106.97, 120.20, 122.03, 124.90, 125.21, 126.68, 127.72, 128.50, 130.71, 133.67, 135.71, 138.68, 144.50, 151.32, 153.66

2.3.16 *cis*-6-(2,4-dimethylphenyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (4n)

Pale white crystals, m.p. 132-134 °C. FTIR (ATR, cm^{-1}): 3379, 3051, 3006, 2919, 2850, 1614, 1573, 1517, 1456, 1400, 1375, 1356, 1290, 1240, 1141, 1106, 1032, 812, 787, 752, 729. ^1H NMR (CDCl_3 , 500 MHz): 1.83-1.84 (dd, $J_1=8.50$ Hz, $J_2=7.00$ Hz, 1H), 2.36 (s, 3H), 2.38 (s, 3H), 2.76-2.82 (dt, $J_1=8.50$ Hz, $J_2=2.00$ Hz, 1H), 3.11-3.17 (ddd, $J_1=8.50$ Hz, $J_2=6.50$ Hz, $J_3=2.00$ Hz, 1H), 4.27-4.28 (d, $J=8.50$ Hz, 1H), 4.36 (br s, 1H), 4.91-4.92 (d, $J=2.00$ Hz, 1H), 5.67 (br s, 1H), 5.93 (br s, 1H), 7.05 (br s, 1H), 7.12-7.14 (d, $J=8.00$ Hz, 1H), 7.21-7.23 (d, $J=8.50$ Hz, 1H), 7.27-7.29 (d, $J=8.50$ Hz, 1H), 7.37-7.42 (m, 2H), 7.69-7.76 (m, 3H). ^{13}C NMR (CDCl_3 , 125 MHz): 19.11, 21.01, 31.47, 43.27, 47.24, 54.27, 118.64, 119.55, 120.27, 123.66, 125.07, 125.12, 126.16, 126.96, 127.38, 128.64, 130.83, 131.38, 132.59, 134.05, 134.96, 136.57, 137.86, 140.53. MS m/z (ESI): 326 (M^+), 261, 222, 207, 171.

2.3.17 *cis*-6-(2,4-dichlorophenyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (4o)

Pale yellow crystals, m.p. 80-82 °C. FTIR (ATR, cm^{-1}): 3367, 3052, 3030, 2949, 2928, 2868, 2841, 1589, 1575, 1560, 1517, 1468, 1401, 1383, 1354, 1290, 1267, 1240, 1196, 1145, 1100, 1048, 1034, 864, 814, 790, 752, 731. ^1H NMR (CDCl_3 , 500 MHz): 1.74-1.79 (dd, $J_1=8.50$ Hz, $J_2=6.50$ Hz, 1H), 2.58-2.64 (dt, $J_1=8.50$ Hz, $J_2=2.50$ Hz, 1H), 3.25-3.30 (ddd, $J_1=8.50$ Hz, $J_2=6.50$ Hz, $J_3=2.50$ Hz, 1H), 4.22 (br s, 1H), 4.24-4.26 (d, $J=8.50$ Hz, 1H), 5.06-5.07 (d, $J=2.50$ Hz, 1H), 5.23 (br s, 1H), 5.62 (br s, 1H), 5.93 (br s, 1H), 7.17-7.19 (d, $J=8.00$ Hz,

1H), 7.27-7.29 (d, $J= 8.50$ Hz, 1H), 7.32-7.34 (dd, $J_1= 8.50$ Hz, $J_2= 2.00$ Hz, 1H), 7.36-7.42 (m, 3H), 7.69-7.71 (d, $J= 8.00$ Hz, 1H), 7.72-7.74 (dd, $J_1= 8.50$ Hz, $J_2= 2.00$ Hz, 1H), 7.77-7.79 (d, $J= 8.50$ Hz, 1H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.35, 42.09, 46.70, 54.33, 119.37, 119.45, 120.44, 123.83, 125.27, 127.31, 127.46, 128.78, 128.94, 129.57, 130.46, 132.63, 133.50, 134.15, 138.87, 139.40. MS m/z (ESI): 366 (M^+), 300, 245, 171.

2.3.18 *trans*-6-(2,4-dichlorophenyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (**4o'**)

Pale yellow oil. ^1H NMR (CDCl_3 , 500 MHz): 1.99-2.01 (d, $J= 10.00$ Hz, 1H), 2.38-2.42 (m, 1H), 2.72-2.76 (m, 1H), 2.95-2.97 (d, $J= 8.50$ Hz, 1H), 3.25 (br s, 1H), 4.47-4.49 (d, $J= 8.00$ Hz, 1H), 4.59 (br s, 1H), 5.50 (br s, 1H), 5.70 (br s, 1H), 7.15-7.17 (d, $J= 8.50$ Hz, 1H), 7.30-7.32 (d, $J= 8.50$ Hz, 1H), 7.37-7.42 (m, 4H), 7.53-7.55 (d, $J= 8.50$ Hz, 1H), 7.61-7.63 (m, 1H), 7.75-7.78 (m, 1H). ^{13}C NMR (CDCl_3 , 125 MHz): 37.20, 43.40, 48.05, 55.15, 118.70, 119.03, 121.90, 122.93, 125.08, 126.76, 127.15, 127.62, 128.65, 129.20, 130.03, 132.45, 133.50, 134.40, 140.53, 141.02

2.3.19 *cis*-6-(2,4-difluorophenyl)-6,6a,7,9a-tetrahydro-5H-benzo[h]cyclopenta[c]quinoline (**4p**)

White crystals, m.p. 128-130 °C. FTIR (ATR, cm^{-1}): 3359, 3064, 3030, 2924, 2904, 2848, 1614, 1595, 1573, 1517, 1498, 1463, 1425, 1398, 1375, 1356, 1293, 1271, 1246, 1143, 1129, 1091, 1034, 956, 856, 849, 818, 791, 757, 739. ^1H NMR (CDCl_3 , 500 MHz): 1.85-1.90 (dd, $J_1= 8.50$ Hz, $J_2= 7.00$ Hz, 1H), 2.64-2.69 (dt, $J_1= 8.50$ Hz, $J_2= 2.50$ Hz, 1H), 3.17-3.23 (ddd, $J_1= 8.50$ Hz, $J_2= 6.00$ Hz, $J_3= 2.50$ Hz, 1H), 4.27-4.29 (d, $J= 8.50$ Hz, 1H), 5.03-5.04 (d, $J= 2.50$ Hz, 1H), 5.66 (br s, 1H), 5.95 (br s, 1H), 6.84-6.88 (t, $J= 8.50$ Hz, 1H), 6.97-7.00 (t, $J= 8.50$ Hz, 1H), 7.20-7.22 (d, $J= 8.50$ Hz, 1H), 7.30-7.32 (d, $J= 8.50$ Hz, 1H), 7.39-7.45 (m, 2H), 7.74-7.81 (m, 3H). ^{13}C NMR (CDCl_3 , 125 MHz): 31.45, 43.53, 46.62, 50.71, 119.24, 119.37, 120.34, 123.73, 125.20, 127.27, 128.70, 130.39, 132.53, 134.12, 139.43, 158.79, 161.90, 163.63. MS m/z (ESI): 334 (M^+), 269, 222, 207, 171.

2.3.20 *cis*-6-phenyl-5,6,6a,7,10,10a-hexahydrobenzo[c]phenanthridine (**6a**)

Pale yellow oil. FTIR (ATR, cm^{-1}): 3374, 3050, 3027, 2915, 2862, 1600, 1582, 1512, 1490, 1462, 1400, 1370, 1345, 1282, 1271, 1230, 1137, 1092, 1080, 1019, 815, 776, 752, 731. ^1H NMR (CDCl_3 , 500 MHz): 1.45-1.47 (m, 1H), 1.87-1.95 (m, 1H), 1.97-1.99 (m, 1H), 2.21-2.25 (m, 1H), 3.98 (br s, 1H), 4.55 (br s, 1H), 4.82 (br s, 1H), 5.83 (br s, 1H), 6.32 (br s, 1H), 7.28-7.32 (m, 1H), 7.34-7.39 (m, 1H), 7.41-7.47 (m, 5H), 7.57-7.59 (d, $J= 7.50$ Hz, 2H), 7.75-7.80 (m, 2H). ^{13}C NMR (CDCl_3 , 125 MHz): 17.22, 25.45, 37.57, 39.05, 59.88, 117.99, 118.64, 119.74, 125.00, 125.15, 126.35, 126.77, 127.13, 127.39, 128.45, 128.52, 128.62, 129.55, 132.54, 132.62, 133.17, 133.82, 134.90, 137.36, 137.82, 140.97, 142.32. MS m/z (ESI): 312 (M^+), 233, 184, 156.

2.3.21 *cis*-4-(-5,6,6a,7,10,10a-hexahydrobenzo[c]phenanthridin-6-yl)benzotrile (**6b**)

Pale yellow oil. FTIR (ATR, cm^{-1}): 3368, 3055, 3007, 2951, 2910, 2860, 2837, 1604, 1570, 1530, 1505, 1480, 1440, 1360, 1290, 1250, 1155, 1092, 1020, 856, 815, 792, 752. ^1H NMR

(CDCl₃, 500 MHz): 1.40-1.45 (m, 1H), 2.27-2.32 (m, 1H), 2.70-2.77 (m, 1H), 3.06-3.13 (m, 1H), 3.40 (br s, 1H), 4.70 (br s, 1H), 5.30 (br s, 1H), 5.70 (br s, 1H), 6.54 (br s, 1H), 7.17-7.22 (m, 1H), 7.37-7.40 (m, 1H), 7.43-7.49 (m, 2H), 7.52-7.56 (m, 2H), 7.69-7.76 (m, 2H), 7.80-7.85 (m, 2H). ¹³C NMR (CDCl₃, 125 MHz): 23.94, 29.71, 39.30, 40.68, 56.17, 109.26, 115.43, 118.40, 121.76, 123.25, 123.98, 125.94, 130.97, 137.72, 141.89, 142.54. MS *m/z* (ESI): 337 (M⁺), 285, 260, 184.

2.3.22 *cis*-6-(4-chlorophenyl)-5,6,6a,7,10,10a-hexahydrobenzo[*c*]phenanthridine (6c)

Pale yellow oil. FTIR (ATR, cm⁻¹): 3385, 3060, 3004, 2947, 2928, 2840, 1575, 1524, 1485, 1468, 1385, 1375, 1351, 1284, 1270, 1240, 1126, 1082, 1012, 823, 787, 770, 753, 736. ¹H NMR (CDCl₃, 500 MHz): 1.80-1.82 (m, 1H), 1.84-1.87 (m, 1H), 1.90-1.93 (m, 1H), 2.10-2.13 (m, 1H), 3.89 (br s, 1H), 4.39 (br s, 1H), 4.72 (br s, 1H), 5.75 (br s, 1H), 6.24 (br s, 1H), 7.09-7.12 (d, *J* = 7.50 Hz, 1H), 7.21-7.23 (d, *J* = 8.50 Hz, 1H), 7.33-7.37 (m, 4H), 7.44-7.46 (d, *J* = 8.50 Hz, 2H), 7.69-7.72 (t, *J* = 7.50 Hz, 2H). ¹³C NMR (CDCl₃, 125 MHz): 16.11, 25.43, 28.68, 36.44, 52.20, 117.30, 118.63, 119.14, 123.81, 124.08, 124.19, 125.66, 125.88, 126.48, 127.24, 127.45, 127.56, 128.02, 129.68, 130.02, 132.08, 133.26, 134.04, 134.95, 136.48, 137.68, 138.11, 139.79. MS *m/z* (ESI): 346 (M⁺), 267, 185, 157.

2.3.23 *cis*-6-(2,4-dichlorophenyl)-5,6,6a,7,10,10a-hexahydrobenzo[*c*]phenanthridine (6d)

Pale yellow oil. FTIR (ATR, cm⁻¹): 3385, 3073, 3058, 2949, 2917, 2875, 1584, 1573, 1562, 1516, 1471, 1456, 1397, 1339, 1281, 1221, 1116, 1098, 1043, 856, 801, 790, 753, 731. ¹H NMR (CDCl₃, 500 MHz): 1.90-1.92 (m, 1H), 1.94-1.96 (m, 1H), 2.27-2.29 (m, 1H), 2.39-2.41 (m, 1H), 3.98 (br s, 1H), 4.31 (br s, 1H), 5.12 (br s, 1H), 5.84 (br s, 1H), 6.33 (br s, 1H), 7.30-7.32 (d, *J* = 8.50 Hz, 1H), 7.37-7.39 (d, *J* = 8.50 Hz, 1H), 7.41-7.44 (m, 2H), 7.48-7.54 (m, 2H), 7.76-7.78 (d, *J* = 8.50 Hz, 2H), 7.84-7.86 (d, *J* = 8.50 Hz, 1H). ¹³C NMR (CDCl₃, 125 MHz): 17.53, 25.37, 29.71, 37.19, 56.19, 115.09, 118.65, 119.60, 120.90, 121.13, 122.41, 123.37, 123.90, 124.78, 125.18, 126.72, 126.93, 128.44, 128.59, 129.15, 129.72, 130.95, 132.59, 133.28, 135.22, 137.86, 140.05, 141.50. MS *m/z* (ESI): 381 (M⁺), 302, 220, 184, 156.

3. Results and Discussions

Although 2,3-dihydrofuran and 3,4-dihydro-2H-pyran have been extensively used in three-component aza-Diels-Alder reactions, there are very limited examples of tetrahydroquinolines and hexahydrophenanthridines synthesized by means of the aza-Diels-Alder reactions of cyclopentadiene and 1,3-cyclohexadiene.

In this work, we described a mild and efficient approach for the synthesis of tetrahydroquinoline and hexahydrophenanthridine derivatives via an aza-Diels-Alder reaction using ytterbium (III) triflate as the catalyst with moderate to good yields (Scheme 1, Scheme 2, Table 1 and Table 2). The reaction was first explored by stirring a mixture of 1-naphthylamine, benzaldehyde and cyclopentadiene with 10 mol% of ytterbium (III) triflate at room temperature in CH₃CN for 24 h (Table 3). The N-arylimine was generated in situ by the interaction of 1-naphthylamine with benzaldehyde but the tetrahydroquinoline product was not observed. Then, the same reaction was performed in refluxing CH₃CN for 24 h but again the

tetrahydroquinoline product was not observed. Then, we changed the solvent to PhMe. The tetrahydroquinoline product was not observed at room temperature. When the reaction was performed in refluxing PhMe, the tetrahydroquinoline product was observed with a 55% yield after 6 h (Table 3). The results showed that the reaction temperature had a significant effect on the reaction and further experiments were carried out at 110 °C in PhMe.

Finally, we also performed the same experiments under ultrasonic irradiation in order to observe the effect of the ultrasonic irradiation. 1-naphtylamine, benzaldehyde and cyclopentadiene, with 10 mol% of ytterbium (III) triflate at room temperature in PhMe for 4h did not give the corresponding tetrahydroquinoline. As shown in Table 3, changing the reaction temperature from room temperature to 50 °C gave the tetrahydroquinoline product a 73% yield and the reaction was completed in 40 min.

In order to observe the effect of the amount of ytterbium (III) triflate on the reaction, we also performed the experiments using different amounts of catalyst. As shown in Table 3, when the amount of ytterbium (III) triflate decreased to 10%, 5% and 2%, the yield of the corresponding tetrahydroquinoline increased to 73%, 75% and 79%, respectively.

From the results above, it has been shown that 1-naphtylamine, benzaldehyde and cyclopentadiene, with 2 mol% of ytterbium (III) triflate at 50 °C in PhMe under ultrasonic irradiation presented an efficient procedure in terms of high yield and shorter reaction time.

With these optimal reaction conditions, we then examined a variety of aromatic and hetero-aromatic aldehydes in conventional and ultrasound-promoted catalytic aza-Diels-Alder reactions. Several N-arylimines (formed in situ from aromatic aldehydes and 1-naphtylamine in PhMe) reacted smoothly with cyclopentadiene under conventional and ultrasonic techniques to afford the corresponding tetrahydroquinolines. In every case ultrasonic irradiation improved the yields and the reactions were completed in 40 minutes (Table 1). The ultrasonic method was equally effective for both electron-rich and electron-deficient aromatic aldehydes. Furthermore, thienyl, furanyl and pyridinyl rings worked well without any decomposition under these reaction conditions.

We also performed the catalytic aza-Diels-Alder reaction of 1-naphtylamine, aromatic aldehydes and 1,3-cyclohexadiene under conventional and ultrasonic techniques. The reactions proceeded smoothly to give the hexahydrophenathridine derivatives with moderate yields (Table 2).

The products were isolated from the reaction mixture by chromatographic purification and identified by FTIR, ¹H NMR, ¹³C NMR and mass spectroscopic data. Each of the products proved to be a pure, single diastereoisomer, by their NMR spectra.

COSY and NOE studies were carried out to determine the absolute configuration of the newly generated stereogenic centre of the products. In most cases, the product was obtained as a single *cis*-isomer. The aza-Diels-Alder reaction of cyclopentadiene, 1-naphtylamine and aromatic aldehydes (4-chlorobenzaldehyde, 2,4-dichlorobenzaldehyde and 5-methyl-2-furan carboxaldehyde) gave the corresponding tetrahydroquinolines as a mixture of *cis*- and *trans*-

isomers (**4c-4c'**, **4m-4m'**, **4o-4o'**), while favoring the *cis*-isomer. The isomers could be easily separated by column chromatography on silica gel. The ratio of the isomers was determined from ¹H NMR spectra of the crude products. Using conventional methods the obtained ratios of *cis*- and *trans*- isomers were 98:2, 98:2 and 93:7, respectively. Under ultrasonic irradiation, the ratios of *cis*- and *trans*- isomers were 92:8, 80:20 and 90:10, respectively. The stereoconfigurations of **4c** and **4c'** were determined by COSY and NOE experiments (Fig 1). The ¹H-¹H-COSY spectra of compound **4c** showed cross peaks between H₂ and H₃, and among H₄, H_{7a}, H₃ and H₅, respectively. The NOE experiment of compound **4c** showed correlations among H₄, H₂, H₃ and H₅. The ¹H-¹H-COSY spectra of compound **4c'** showed cross peaks among H₂, H₃, H₆ and H_{7a}; and between H₄ and H₃, respectively. The NOE experiment of compound **4c'** showed correlation between H₃ and H₄.

4. Conclusion

In conclusion, we have described a simple and efficient method for the synthesis of tetrahydroquinolines and hexahydrophenathridines via a three-component one-pot aza-Diels-Alder reaction of 1-naphtylamine, aromatic aldehydes and cyclic dienes using commercially available ytterbium (III) trifluoromethanesulfonate (Yb(OTf)₃) as the catalyst.

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References

- [1] (a) A. Dömling, Recent developments in isocyanide based multicomponent reactions in applied chemistry, *Chem. Rev.* 106 (2006) 17-89;
- (b) J. Zhu, Recent developments in the isonitrile-based multicomponent synthesis of heterocycles, *Eur. J. Org. Chem.* (2003) 1133-1144;
- (c) J. Zhu, H. Bienayme, *Multicomponent Reactions*, Eds.; Wiley-VCH, Weinheim-Germany, 2005;
- (d) J. D. Sunderhaus, S. F. Martin, Applications of multicomponent reactions to the synthesis of diverse heterocyclic scaffolds, *Chem.-Eur. J.* 15 (2009) 1300-1308;
- [2] (a) S. Periyaraja, P. Shanmugam, A. B. Mandal, T.S. Kumar, P. Ramamurthy, Unusual reactivity of 1-aminoanthraquinone in copper catalyzed multicomponent reaction with isatins and aryl alkynes: synthesis and photophysical properties of regioisomeric fluorescent 3-spiroheterocyclic 2-oxindoles, *Tetrahedron* 69 (2013) 2891-2899;
- (b) V. Sridharan, K. Karthikeyan, S. Muthusubramanian, Unexpected multicomponent reaction of 2/4-methoxyarylaldehydes with arylhydroxylamines and maleic anhydride: a novel synthesis of unsymmetrical diarylamines, *Tetrahedron Lett.* 47 (2006) 4221-4223;
- (c) S.L. Zhu S.J. Ji X.M. Su C. Sun Y. Liu, Facile and efficient synthesis of a new class of bis(3'-indolyl)pyridine derivatives via one-pot multicomponent reactions, *Tetrahedron Lett.* 49 (2008) 1777-1781;

- [3] (a) M.Z. Hoemann, R.L. Xie, R.F. Rossi, S. Meyer, A. Sidhu, G.D. Cuny, J.R. Hauske, Potent *in vitro* methicillin-resistant *Staphylococcus aureus* activity of 2-(1*H*-indol-3-yl)tetrahydroquinoline derivatives, *Bioorg. & Med. Chem. Lett.* 12 (2002) 129-132;
- (b) N. Nagata, M. Miyakawa, S. Amano, K. Furuya, N. Yamamoto, K. Inoguchi, Design and synthesis of tricyclic tetrahydroquinolines as a new series of nonsteroidal selective androgen receptor modulators (SARMs), *Bioorg. & Med. Chem. Lett.* 21 (2011) 1744-1747;
- (c) D.S. Su, J.J. Lim, E. Tinney, B.L. Wana, M.B. Young, K.D. Anderson, D. Rudd, V. Munshi, C. Bahnck, P.J. Felock, M. Lu, M.T. Lai, S. Touch, G. Moyer, D.J. DiStefano, J.A. Flynn, Y. Liang, R. Sanchez, S. Prasad, Y. Yan, R. Perlow-Poehnelt, M. Torrent, M. Miller, J.P. Vacca, T.M. Williams, N.J. Anthony, Substituted tetrahydroquinolines as potent allosteric inhibitors of reverse transcriptase and its key mutants, *Bioorg. & Med. Chem. Lett.* 19 (2009) 5119-5123;
- (d) V.K. Gore, V.V. Ma, R. Yin, J. Ligutti, D. Immke, E.M. Doherty, M.H. Norman, Structure-activity relationship (SAR) investigations of tetrahydroquinolines as BKCa agonists, *Bioorg. & Med. Chem. Lett.* 20 (2010) 3573-3578;
- (e) K. Kavr, M. Jain, R.P. Reddy, R. Jain, Quinolines and structurally related heterocycles as antimalarials, *Eur. J. Med. Chem.* 45 (2010) 3245-3264;
- (f) A. Munoz, F. Soja, D.R.M. Arenas, V.V. Kouznetsov, F. Arvelo, Cytotoxic effects of new *trans* 2,4-diaryl-*r*-3-methyl-1,2,3,4-tetrahydroquinolines and their interaction with antitumoral drugs gemcitabine and paclitaxel on cellular lines of human breast cancer, *Chem.-Bio. Interactions* 189 (2011) 215-221;
- (g) H.M. Faidallah, S.A.F. Rostom, Synthesis, *in vitro* antitumor evaluation and DNA-binding study of novel tetrahydroquinolines and some derived tricyclic and tetracyclic ring systems, *Eur. J. Med. Chem.* 63 (2013) 133-143;
- (h) A.L. Ruchelman, S. Zhu, . Zhou, A. Liu, L.F. Liu, E.J. La Voie, Dimethoxybenzo[*i*]phenanthridine-12-carboxylic acid derivatives and 6*H*-dibenzo[*c,h*][2,6]naphthyridin-5-ones with potent topoisomerase I-targeting activity and cytotoxicity, *Bioorg. & Med. Chem. Lett.* 14 (2004) 5585-5589;
- (i) J.P. Cueva, B.R. Chemel, J.I. Juncosa Jr. M.A. Lill, V.J. Watts, D.E. Nichols, Analogues of doxorubicin reveal differences between the dopamine D₁ receptor binding properties of chromanoisoquinolines and hexahydrobenzo[*a*]phenanthridines, *Eur. J. Med. Chem.* 48 (2012) 97-107;
- (j) S.A. Baechler, M. Fehr, M. Habermeyer, A. Hofmann, K.H. Merz, H.H. Fiebig, D. Marko, G. Eisenbrand, Synthesis, topoisomerase-targeting activity and growth inhibition of lycobetaine analogs, *Bioorg. & Med. Chem.* 21 (2013) 814-823;

- [4] (a) A.R. Katritzky, S. Rachwal, B. Rachwal, Recent progress in the synthesis of 1,2,3,4,-tetrahydroquinolines, *Tetrahedron* 52 (1996) 15031-15070;
- (b) R. Perez-Ruiz, L.R. Domingo, M.C. Jimenez, M.A. Miranda, Experimental and theoretical studies on the radical-cation-mediated imino-Diels–Alder reaction, *Org. Lett.* 13 (2011), 5116-5119;
- (c) V.V. Kouznetsov, Recent synthetic developments in a powerful imino Diels–Alder reaction (Povarov reaction): application to the synthesis of *N*-polyheterocycles and related alkaloids, *Tetrahedron* 65 (2009) 2721-2750;
- (d) P. Buonora, J.C. Olsen, T. Oh, Recent developments in imino Diels-Alder reactions, *Tetrahedron* 57 (2001) 6099-6138;
- [5] F. Fadel, S.L. Titouani, M. Soufiaoui, H. Ajamay, A. Mazzah, Synthèse de nouveaux dérivés tétrahydroquinoléines et quinoléines via la réaction d'aza-Diels-Alder suivie d'aromatization, *Tetrahedron Lett.* 45 (2004) 5905-5908;
- [6] J.F. Jr. Kervin, S. Danishefsky, On the Lewis acid catalyzed cyclocondensation of imines with a siloxydiene *Tetrahedron Lett.* 23 (1982) 3739-3742;
- [7] W. Zhang, Y. Dai, X. Wang, W. Zhang, One-pot synthesis of pyrrolidino- and piperidinoquinolines by three-component aza-Diels-Alder reactions of in situ generated *N*-arylimines and cyclic enamides, *Tetrahedron Lett.* 52 (2011) 6122-6126;
- [8] K. Makino, Y. Henmi, M. Terasawa, O. Hara, Y. Hamada, Remarkable effects of titanium tetrachloride in diastereoselective aza Diels–Alder cycloaddition: synthesis of (*S*)-piperazic acid, *Tetrahedron Lett.* 46 (2005) 555-558;
- [9] (a) G. Babu, P.T. Perumal, Imino Diels-Alder reactions catalyzed by indium trichloride (InCl_3): Facile synthesis of quinoline and phenanthridinone derivatives, *Tetrahedron Lett.* 38 (1997) 5025-5026;
- (b) J. Zhang, C-J. Li, InCl_3 catalyzed domino reaction of aromatic amines with cyclic eno ethers in water: A highly efficient synthesis of new 1,2,3,4-tetrahydroquinoline derivatives, *J. Org. Chem.* 67 (2002) 3969-3971;
- [10] (a) B. Das, M.R. Reedy, V.S. Reedy, R. Ramu, Novel and efficient Lewis acids as catalysts for single-step synthesis of pyrano- and furoquinolines, *Chem. Lett.* 33 (2004) 1526-1528;
- (b) M. Mahesh, Ch. Venkateshwar Reddy, K. Srinivasa Reddy, P.V.K. Raju, V.V. Narayana Reddy, Imino Diels-Alder reactions: Efficient synthesis of pyrano and furoquinolines catalyzed by ZrCl_4 , *Synth. Commun.* 34 (2004) 4089-4105;
- [11] (a) P.A. Grieco, A. Bahsas, Role reversal in the cyclocondensation of cyclopentadiene with heterodienophiles derived from aryl amines and aldehydes: Synthesis of novel tetrahydroquinolines, *Tetrahedron Lett.* 29 (1988) 5855-5858;

- (b) J.M. Mellor, G.D. Merriman, P. Riviere, Synthesis of tetrahydroquinolines from aromatic amines, formaldehyde and electron rich alkenes: Evidence for nonconcertedness, *Tetrahedron* 32 (1991) 7103-7106;
- [12] D.L. Boger, In *Comprehensive Organic Synthesis*, B.M. Trost, I. Fleming, Eds., Pergamon: Oxford, 1991, Vol 5, p 451;
- [13] T. Akiyama, J. Takaya, H. Kagoshima, Brønsted acid-catalyzed aza Diels-Alder reaction of Danishefsky's diene with aldimine generated in situ from aldehyde and amine in aqueous media, *Tetrahedron Lett.* 40 (1999) 7831-7834;
- [14] (a) S. Kobayashi, M. Araki, H. Ishitani, S. Nagayama, I. Hachiya, Activation of Imines by Rare Earth Metal Triflates. Ln(OTf)₃- or Sc(OTf)₃-Catalyzed Reactions of Imines with Silyl Enolates and Diels-Alder Reactions of Imines, *Synlett* 3 (1995) 233-234;
- (b) S. Luo, L. Zhu, A. Taludar, G. Zhang, X. Mi, J. Chjeng, P. Wang, Recent advances in rare earth-metal triflate catalyzed organic synthesis in green media, *Mini-Reviews in Organic Chemistry* 2 (2005) 546-564;
- (c) K. Turhan, E. Pelit, Z. Turgut, Aza-Diels–Alder reactions with lanthanide triflates: syntheses of quinoline and phenanthridine derivatives, *Synt. Commun.* 39 (2009) 1729-1741;
- [15] (a) V. Sridharan, C. Avendano, J.C. Menendez, CAN-catalyzed three-component reaction between anilines and alkyl vinyl ethers: Stereoselective synthesis of 2-methyl-1,2,3,4-tetrahydroquinolines and studies on their aromatization, *Tetrahedron* 63 (2007) 673-681;
- (b) G. Savitha, P.T. Perumal, An efficient one-pot synthesis of tetrahydroquinoline derivatives via an aza Diels-Alder reaction mediated by CAN in an aqueous medium and oxidation to heteroarylquinolines, *Tetrahedron Lett.* 47 (2006) 3589-3593;
- [16] E. Rajanendar, M.N. Reddy, K.G. Reddy, S.R. Krishna, L-Proline catalyzed efficient one-pot three-component aza-Diels–Alder reactions on nitrostyrylisoxazoles: a facile synthesis of new isoxazolyl tetrahydroquinolines and isoxazolo[2,3-*a*]pyrimidines, *Tetrahedron Lett.* 53 (2012) 2909-2913;
- [17] K. Nagaiah, D. Sreenu, R. Srinivasa Rao, G. Vashishta, J.S. Yadav, Phosphomolybdic acid-catalyzed efficient one-pot three-component aza-Diels–Alder reactions under solvent-free conditions: a facile synthesis of *trans*-fused pyrano- and furanotetrahydroquinolines, *Tetrahedron Lett.* 47 (2006) 4409-4413;
- [18] J.S. Yadav, B.V.S. Reddy, K. Sadasiv, P.S.R. Reddy, Montmorillonite clay-catalyzed [4+2] cycloaddition reactions: A facile synthesis of pyrano- and furanoquinolines, *Tetrahedron Lett.* 43 (2002) 3853-3856;
- [19] S. Palaniappan, B. Rajender, M. Umashankar, Controllable stereoselective synthesis of *cis* or *trans* pyrano and furano tetrahydroquinolines: Polyaniline-*p*-toluenesulfonate salt

- catalyzed one-pot aza-Diels–Alder reactions, *J. Mol. Catalysis A: Chemical*, 352 (2012) 70-74;
- [20] (a) W. Zhang, Y. Guo, Z. Liu, X. Jin, L. Yang, Z.L. Liu, Photochemically catalyzed Diels–Alder reaction of arylimines with *N*-vinylpyrrolidinone and *N*-vinylcarbazole by 2,4,6-triphenylpyrylium salt: synthesis of 4-heterocycle-substituted tetrahydroquinoline derivatives, *Tetrahedron* 61 (2005) 1325–1333;
- (b) X. Xing, J. Wu, W.M. Dai, Acid-mediated three-component aza-Diels–Alder reactions of 2-aminophenols under controlled microwave heating for synthesis of highly functionalized tetrahydroquinolines. Part 9: Chemistry of aminophenols, *Tetrahedron* 62 (2006) 11200-11206;
- (c) L. Chen, C.J. Li, Domino reaction of anilines with 3,4-dihydro-2H-pyran catalyzed by cation-exchange resin in water: an efficient synthesis of 1,2,3,4-tetrahydroquinoline derivatives, *Green Chem.* 5 (2003) 627–629;
- (d) J.S. Yadav, B.V.S. Reddy, J.S.S. Reddy, R. Srinivasa Rao, Aza-Diels–Alder reactions in ionic liquids: a facile synthesis of pyrano- and furanoquinolines, *Tetrahedron* 59 (2003) 1599-1604;
- [21] (a) M. Vinatoru, E. Bartha, F. Badea, J.L. Luche, Sonochemical and thermal redox reactions of triphenylmethane and triphenylmethyl carbinol in nitrobenzene, *Ultrason. Sonochem.* 5 (1998) 27-31;
- (b) M. Meciarova, S. Toma, J.L. Luche, The sonochemical arylation of malonic esters mediated by manganese triacetate, *Ultrason. Sonochem.* 8 (2001) 119-122;
- (c) N. Cabello, P. Cintas, J.L. Luche, Sonochemical effects in the additions of furan to masked *ortho*-benzoquinones, *Ultrason. Sonochem.* 10 (2003) 25-31;
- [22] (a) T. Javed, T.J. Mason, S.S. Phull, N.R. Baker, A. Robertson, Influence of ultrasound on the Diels-Alder cyclization reaction: synthesis of some hydroquinone derivatives and lonapalene, an anti-psoriatic agent, *Ultrason. Sonochem.* 2 (1995) S3-S4;
- (b) J.L. Bravo, I. Lopez, P. Cintas, G. Silvero, M.J. Aravalo, Sonochemical cycloadditions in ionic liquids. Lessons from model cases involving common dienes and carbonyl dienophiles, *Ultrason. Sonochem.* 13(2006) 408-411;
- (c) H. Zheng, H. Li, H. Shao, One-pot three-component Mannich-type reactions using Sulfamic acid catalyst under ultrasound irradiation, *Ultrason. Sonochem.* 16 (2009) 758-762;
- (d) R. Cella, H.A. Stefani, Ultrasound in heterocycles chemistry, *Tetrahedron* 65 (2009) 2619-2641;
- [23] (a) T.J. Mason, D. Peters, *Practical Sonochemistry*, second ed., Power ultrasound uses and applications, Ellis Horwood, New York, 2002;

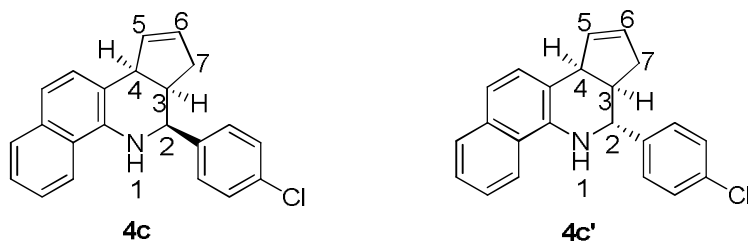
(b) T.J. Mason, Sonochemistry and the environment – Providing a ‘green’ link between chemistry, physics and engineering, *Ultrason. Sonochem.* 14 (2007) 476–483;

[24] (a) J.L. Luche, *Synthetic Organic Sonochemistry*, Plenum Press, New York, 1998;

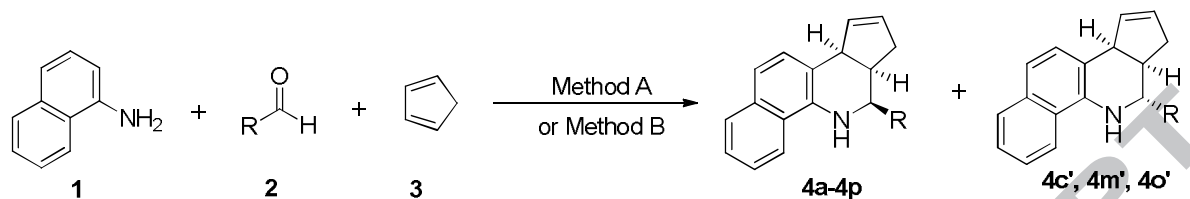
(b) J.T. Li, S.X. Wang, G.F. Chen, T.S. Li, Some applications of ultrasound irradiation in organic synthesis, *Curr. Org. Synth.* 2 (2005) 415-436;

(c) H. Kumar, A. Parmar, Ultrasound promoted $ZrCl_4$ catalyzed rapid synthesis of substituted 1,2,3,4-tetrahydropyrimidine-2-ones in solvent or dry media, *Ultrason. Sonochem.* 15 (2008) 129-132;

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Figure 1. Compounds **4c** and **4c'**

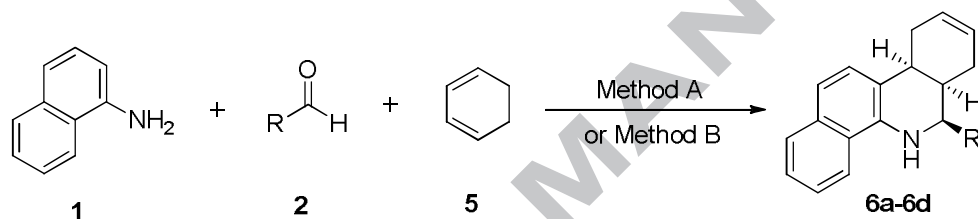
Scheme 1. One-pot aza-Diels-Alder reaction of amine, aldehydes and cyclopentadiene catalyzed by $\text{Yb}(\text{OTf})_3$



Method A: 10 mol% $\text{Yb}(\text{OTf})_3$, reflux in PhMe at 110 °C for 6h.

Method B: 2 mol% $\text{Yb}(\text{OTf})_3$, ultrasound in PhMe at 50 °C for 40 min.

Scheme 2. One-pot aza-Diels-Alder reaction of amine, aldehydes and 1,3-cyclohexadiene catalyzed by $\text{Yb}(\text{OTf})_3$



Method A: 10 mol% $\text{Yb}(\text{OTf})_3$, reflux in PhMe at 110 °C for 6h.

Method B: 2 mol% $\text{Yb}(\text{OTf})_3$, ultrasound in PhMe at 50 °C for 40 min.

Table 1. Direct aza-Diels-Alder reaction with cyclopentadiene

Entry	R	Compound	% Yield (Method A) ^{a,c}	% Yield (Method B) ^{b,c}
1	Phenyl	4a	55	79
2	4-Bromophenyl	4b	54	76
3	4-Chlorophenyl	4c+4c'	56 ^d	82 ^d
4	4-Methylphenyl	4d	55	78
5	4-Methoxyphenyl	4e	49	72
6	4-Hydroxyphenyl	4f	52	83
7	4-Cyanophenyl	4g	58	79
8	4-Nitrophenyl	4h	47	74
9	2-Nitrophenyl	4i	50	78
10	2-Pyridinyl	4j	46	68
11	2-Thiophenyl	4k	43	57
12	3-Furanyl	4l	52	77
13	5-Methyl-2-furanyl	4m+4m'	45 ^d	67 ^d
14	2,4-Dimethylphenyl	4n	65	81
15	2,4-Dichlorophenyl	4o+4o'	53 ^d	65 ^d
16	2,4-Difluorophenyl	4p	67	86

^aMethod A: 1-naphthylamine (1 mmol), aromatic aldehyde (1 mmol), cyclopentadiene (3 mmol), 10 mol% Yb(OTf)₃, dry toluene (3 mL), at 110 °C, 6 h.

^bMethod B: 1-naphthylamine (1 mmol), aromatic aldehyde (1 mmol), cyclopentadiene (3 mmol), 2 mol% Yb(OTf)₃, dry toluene (3 mL), ultrasonic irradiation, at 50 °C, 40 min.

^cIsolated yield.

^dCombined yields of two isomer.

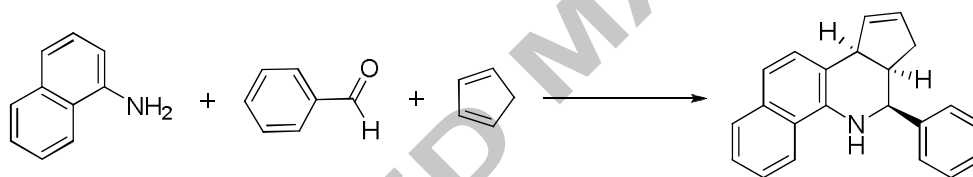
Table 2. Direct aza-Diels-Alder reaction with 1,3-cyclohexadiene

Entry	R	Compound	Yield % (Method A) ^{a,c}	Yield % (Method B) ^{b,c}
1	Phenyl	6a	15	27
2	4-Cyanophenyl	6b	12	21
3	4-Chlorophenyl	6c	14	24
4	2,4-Dichlorophenyl	6d	10	16

^aMethod A: 1-naphthylamine (1 mmol), aromatic aldehyde (1 mmol), 1,3-cyclohexadiene (3 mmol), 10 mol% Yb(OTf)₃, dry toluene (3 mL), at 110 °C, 6 h.

^bMethod B: 1-naphthylamine (1 mmol), aromatic aldehyde (1 mmol), 1,3-cyclohexadiene (3 mmol), 2 mol% Yb(OTf)₃, dry toluene (3 mL), ultrasonic irradiation, at 50 °C, 40 min.

^cIsolated yield.

Table 3. The direct aza-Diels-Alder reaction^a

Entry	Method	Solvent	Catalyst	Temperature (°C)	Time	Yield (%) ^b
1	Stirring	CH ₃ CN	10% Yb(OTf) ₃	r.t	24 h	-
2	Refluxing	CH ₃ CN	10% Yb(OTf) ₃	80	24 h	-
3	Stirring	PhMe	10% Yb(OTf) ₃	r.t	24 h	-
4	Refluxing	PhMe	10% Yb(OTf) ₃	110	6 h	55
5	Ultrasound	PhMe	10% Yb(OTf) ₃	r.t	4 h	-
6	Ultrasound	PhMe	10% Yb(OTf) ₃	50	40 min	73
7	Ultrasound	PhMe	5% Yb(OTf) ₃	50	40 min	75
8	Ultrasound	PhMe	2% Yb(OTf) ₃	50	40 min	79

^aReaction conditions: 1-naphthylamine (1 mmol), benzaldehyde (1 mmol), cyclopentadiene (3 mmol), dry toluene (3 mL)

^bIsolated yield

•Aza-Diels-Alder reaction performed under conventional/ultrasonic techniques •Ultrasonic irradiation provided higher yields and shorter reaction times •Ultrasound-assisted reactions occurred in milder conditions

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